# How to Prepare a Professional Presentation

What to do and what not to do!

# At the Beginning...

- Know to whom you will be speaking
- Know the setting in which you will be speaking
- Know the equipment that will be available

#### Your Audience

- Is your audience primarily faculty, peers, or a mix?
- Are most people from on-campus, local, the region, etc.?
- What is the background of your audience?

## What is the Venue?

- On-campus or off?
- A professional meeting, state capitol, or class?
- What is the purpose of the presentation?
- Is the presentation time fixed?

# What Equipment is Available?

- Oral or poster presentation?
- For posters, what is the permitted size and how close together are the posters? What is the backing?
- For oral presentations, is there an internet connection? Speakers for audio? Sufficient computer speed for movies?

## The Presentation

This is **not** entertainment, but there is no requirement that your bore your audience.

## The Do's

- Your audience is intelligent. Treat it with respect.
- Every slide should have an obvious purpose.
- The font size should be large enough for everyone to read.
- Speak clearly, at a reasonable pace, and make eye contact with the audience.
- Make your slides self-consistent.
- Know what you are talking about.

# The Do's (continued)

- Tell them what you want them to learn from a slide.
- Provide adequate background.
- Take visual cues from the audience.
- If you are asked a question and don't know the answer, admit it.
- Include appropriate acknowledgements and thank appropriately.

# The Do's (continued)

- Know the material well enough to answer questions. You should be the most knowledgeable person in the room on the subject.
- Practice early and often!

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- Cluttered slides
- Including sounds that are not instructional
- Excess or inappropriate humor
- Do not assume the audience will remember everything you say. Judicious repeating can be good.

# The Don'ts (con't)

- No undefined abbreviations unless everyone should know them (think mL = milliliter)
- Don't read to your audience, either off the slides or notecards.
- Nothing to eat or drink while you're presenting. You can make it an hour without rehydrating.

## More Don'ts

 $\blacksquare$  [Na(diglyme)<sub>3/2</sub>][(C<sub>5</sub>Ph<sub>5</sub>)Cr(CO)<sub>3</sub>] (1.00 g, 1.24 mmol) and [PPN]Cl (0.69 g, 1.2 mmol) were each dissolved in 5 mL CH<sub>2</sub>Cl<sub>2</sub>. The solutions were combined, stirred for 20 min, filtered to remove the NaCl precipitate, and dried in vacuo. The resulting yellow powder was stirred with 5 mL acetone for 5 min to dissolve any residual PPN Cl. The acetone was removed by cannula filtration, and the solid was dried in vacuo. The resulting solid was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) and layered with an equal volume of hexanes. The layers were allowed to combine in the dark producing large orange blocks of 2 in 70% yield (0.97 g, 0.87 mmol). Mp: 241-242 °C. Anal. Calcd. for C74H55CrNO3P2: C, 79.34; H, 4.95. Found C, 79.08; H, 5.08.

# and it goes on ...

- $1.00 \text{ g } [\text{Na}(\text{diglyme})_{3/2}][(\text{C}_5\text{Ph}_5)\text{Cr}(\text{CO})_3]$
- 0.69 g [PPN]Cl
- Dissolved separately in 5 mL CH<sub>2</sub>Cl<sub>2</sub>
- The solutions were combined and stirred for 20 min
- Filtered to remove the NaCl.
- Dried *in vacuo* for 1 hour.
- Yellow product extracted with 5 mL acetone for 5.
- Filtered off acetone, and the solid was dried *in vacuo*.
- Dissolved the solid in 15 mL CH<sub>2</sub>Cl<sub>2</sub>
- Solution layered with 15 mL hexanes.
- The tube was allow to sit for 3 days to mix.

# What I actually do.

$$[Na(diglyme)_{3/2}][(C_5Ph_5)Cr(CO)_3] + [PPN]Cl \xrightarrow{THF/CH_2Cl_2} \rightarrow \\ [PPN][(C_5Ph_5)Cr(CO)_3] + NaCl + \frac{3}{2} \text{ diglyme}$$

$$70\% \text{ yield}$$

$$PPN = Ph_3P=N=PPh_3^+$$

$$v(C=O) \quad 1783, 1893 \text{ cm}^{-1} \text{ (CH}_3CN)$$

$$v(C=O) \quad 1792, 1895 \text{ cm}^{-1} \text{ (THF)}$$